# FYS3410 - Vår 2010 (Kondenserte fasers fysikk)

http://www.uio.no/studier/emner/matnat/fys/FYS3410/index-eng.xml
Based on Introduction to Solid State Physics by Kittel

### **Course content**

- Periodic structures, understanding of diffraction experiment and reciprocal lattice
- Crystal binding, elastic strain and waves
- Imperfections in crystals: point defects and diffusion
- Crystal vibrations: phonon heat capacity and thermal conductivity
- Free electron Fermi gas: density of states, Fermi level, and electrical conductivity
- Electrons in periodic potential: energy bands theory classification of metals, semiconductors and insulators
- Semiconductors: band gap, effective masses, charge carrier distributions, doping, pn-junctions
- Metals: Fermi surfaces, temperature dependence of electrical conductivity

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### FYS3410 lecture schedule and exams: Spring 2010

M/18/1/2010: W/20/1/2010:	Introduction and motivation. Periodicity and lattices Index system for crystal planes. Crystal structures	2h 1h	
M/25/1/2010: W/27/1/2010:	Reciprocal space, Laue condition and Ewald construction Brillouin Zones. Interpretation of a diffraction experiment	2h 1h	
M/01/2/2010: W/03/2/2010:	Crystal binding, elastic strain and waves Elastic waves in cubic crystals; defects in crystals	2h 1h	
M/08/2/2010: W/10/2/2010:	Defects in crystals; case study - vacancies Diffusion	2h 1h	
M/15/2/2010: W/17/2/2010:	Crystal vibrations and phonons Crystal vibrations and phonons	2h 1h	
M/22/2/2010: W/24/2/2010:	Lattice heat capacity: Dulong-Petit and Einstein models Phonon density of states (DOS) and Debye model	2h 1h	
M/01/3/2010: W/03/3/2010:	General result for DOS; role of anharmonic interactions Thermal conductivity	2h 1h	
M/08/3/2010: W/10/3/2010:	Free electron Fermi gas in 1D and 3D – ground state Density of states, effect of temperature – FD distribution	2h 1h	
M/15/3/2010: W/17/3/2010:	Heat capacity and thermal conductivity of FEFG Repetition	2h 1h	
22/3/2010:	Mid-term exam		

M/12/4/2010: W/14/4/2010:	Drude model and the idea of energy bands Nearly free electron model; Kronig - Penny model	2h 2h	
M/19/4/2010: W/21/4/2010:	no lectures Empty lattice approximation; number of orbitals in a band	2h	
M/26/4/2010: W/28/4/2010:	Semiconductors, effective mass method, intrinsic carriers Impurity states in semiconductors and carrier statistics	2h 2h	
M/03/5/2010: W/05/5/2010:	p-n junctions and heterojunctions surface structure, surface states, Schottky contacts	2h 2h	
M/10/5/2010: W/12/5/2010:	no lectures no lectures		
W/19/5/2010:	Metals and Fermi surfaces	2h	
W26/5/2010:	Repetition	2h	
27-28/5/2010:	Final Exam (sensor: Prof. Arne Nylandsted Larsen at the Aarhus University Denmark, http://person.au.dk/en/anl@phys.au.dk)	,	

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## FEFG in 3D

$$-\frac{\hbar^2}{2m}\bigg(\frac{\partial^2}{\partial x^2}+\frac{\partial^2}{\partial y^2}+\frac{\partial^2}{\partial z^2}\bigg)\psi_k(\mathbf{r})=\boldsymbol{\epsilon}_k\,\psi_k(\mathbf{r})$$

Invoking <u>periodic boundary condition</u> we get traveling waves as solutions:

$$\psi(x+L,y,z) = \psi(x,y,z)$$

$$\psi_{\mathbf{k}}(\mathbf{r}) = \exp(i\mathbf{k}\cdot\mathbf{r}) ,$$

$$k_x = 0 ; \quad \pm \frac{2\pi}{L} ; \quad \pm \frac{4\pi}{L} ; \dots ,$$

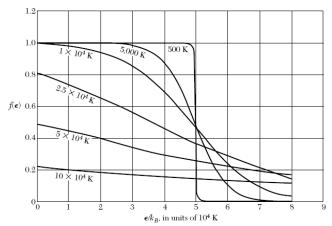
$$\epsilon_{\mathbf{k}} = \frac{\hbar^2}{2m} k^2 = \frac{\hbar^2}{2m} (k_x^2 + k_y^2 + k_z^2)$$
Fermi surface, at energy 
$$\epsilon_F$$

$$k_y$$
4 In the ground state of a system of  $N$  free

Figure 4 In the ground state of a system of N free electrons the occupied orbitals of the system fill a sphere of radius  $k_F$ , where  $\epsilon_F = \hbar^2 k_F^2/2m$  is the energy of an electron having a wavevector  $k_F$ .

## Effect of temperature; Fermi-Dirac distribution

- Describes the probability that an orbit at energy E will be occupied in an ideal electron gas under thermal equilibrium
- $\mu$  is chemical potential, f(  $\varepsilon = \mu$ )=0.5; at 0K,  $\varepsilon_F = \mu$



$$f(\epsilon) = \frac{1}{\exp[(\epsilon - \mu)/k_B T] + 1} \ .$$

Figure 3 Fermi-Dirac distribution function (5) at the various labelled temperatures, for  $T_F \equiv \epsilon_F k_B = 50{,}000$  K. The results apply to a gas in three dimensions. The total number of particles is constant, independent of temperature. The chemical potential  $\mu$  at each temperature may be read off the graph as the energy at which f=0.5.

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#### **Basic assumptions of the Drude model**

### Combining ideas of an electron as a "charge" carrier and kinetic gas theory

- no collisions between electrons independent electron approximation;
- positive charge is located at the immobile ion cores and electronc can collide with the ion cores canging their velocity however in between collisions no interaction is taking place free lectron approximation;
- electrons reach thermal equilibrium with the lattice participating in the collisions so that the mean kinetic energy is 1/2 ( $mv_t^2$ ) =  $3/2(k_BT)$
- $\tau$  is a time in between collisions and  $\lambda = v_t \tau$  is a mean free path
- number of electrons participating in, e.g. conduction, is equivalent to the number of electrons on the outer shell of the atom providing concentrations in the range of 5x10<sup>22</sup> electrons/cm<sup>3</sup>

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$\rm m_e$	k <sub>B</sub>	Т	$v_t^2$	$v_t$
9,10E-31	1,38E-23	300	1,36E+10	1,17E+05
9,10E-31	1,38E-23	273	1,24E+10	1,11E+05
9,10E-31	1,38E-23	77	3,50E+09	5,92E+04
9,10E-31	1,38E-23	15	6,82E+08	2,61E+04

	conductivity	II.	q	lau	Lambua
	1,20E+07	5,00E+28	1,60E-19	8,56E-15	1,00E-09
	1,26E+07	5,00E+28	1,60E-19	8,97E-15	1,00E-09
$-\!\!\!/$	2,38E+07 <b>←</b>	5,00E+28	1,60E-19	1,69E-14	1,00E-09
-	5,38E+07	5,00E+28	1,60E-19	3,83E-14	1,00E-09

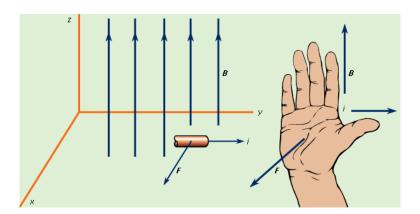
Here we have a problem: the calculated conductivity has increased but the actual experimental increase is significantly bigger.

However, one offthe most convincing pieces of evidence of the Drude theory in his time used to be a qualitatively correct description of the Widemann-Fraz law – the ratio of teermal and electrical conductivity is a constant for all metals at a given temperature:  $\kappa/\sigma$  =LT.

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magnetic force: right-hand rules



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- temperature dependence of the electrical conductivity even for pure metals
   because of unclear of the magnitude of the mean free path
- Scattering of R<sub>H</sub>-values in terms of the sign and magnitudes
- alloying of very small amount of material (doping) may incease the conductivity by several orders of magnitude
- nearly no contribution to the specific heat from electrons at room temperature – contrudiction to the 3R-value by Dulong-Petit.

• ...

!!! FEFG concept explains the last shortcoming. Indeed, only a small fraction of electrons – of the order of kT in the DOS-plot – have to be taken into consideration when calculating the heat capacity related to the electrons.

Bur FEFG can not explain the alloying or Hall effects...

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$$-\frac{\hbar^2}{2m}\bigg(\frac{\partial^2}{\partial x^2}+\frac{\partial^2}{\partial y^2}+\frac{\partial^2}{\partial z^2}\bigg)\psi_{\bf k}({\bf r})=\epsilon_{\bf k}\,\psi_{\bf k}({\bf r})$$
 Adding the the potential created by the rest of the particles around

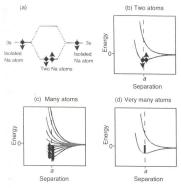
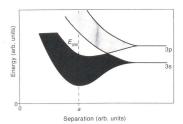


Figure 6.1 The formation of energy bands in solids. (a) Bonding and antibonding energy levels for a cluster of many Na atoms as a function of their separation. (c) The energy levels for a cluster of many Na atoms as a function of their separation. (d) For very many energy levels there is a quasicontinuum between the lowest and highest energy levels. The band is spin. (b) The molecule's energy levels as a



 $\label{eq:Figure 6.2} \textbf{Figure 6.2} \ \ \text{Band formation in Si. The lower band corresponds to} \\ \text{the sp}^3 \ \text{states and is completely filled}.$ 

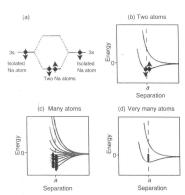


Figure 6.1 The formation of energy bands in solids. (a) Bonding and antibonding energy levels fand their occupation for a molecule constructed from two Na atoms. The black dost and arrows symbolize the electrons with their spin. (b) The molecule's energy levels as a