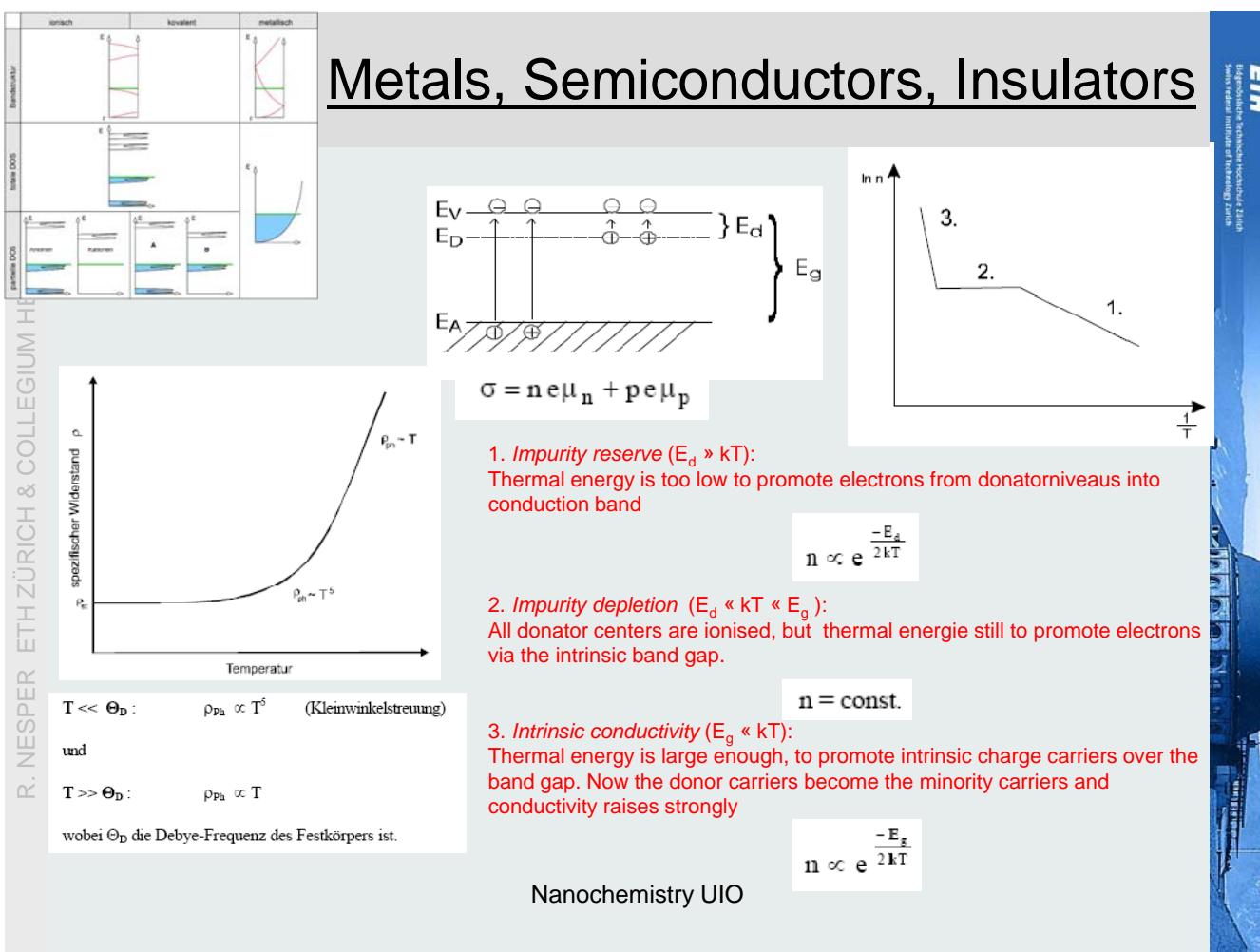
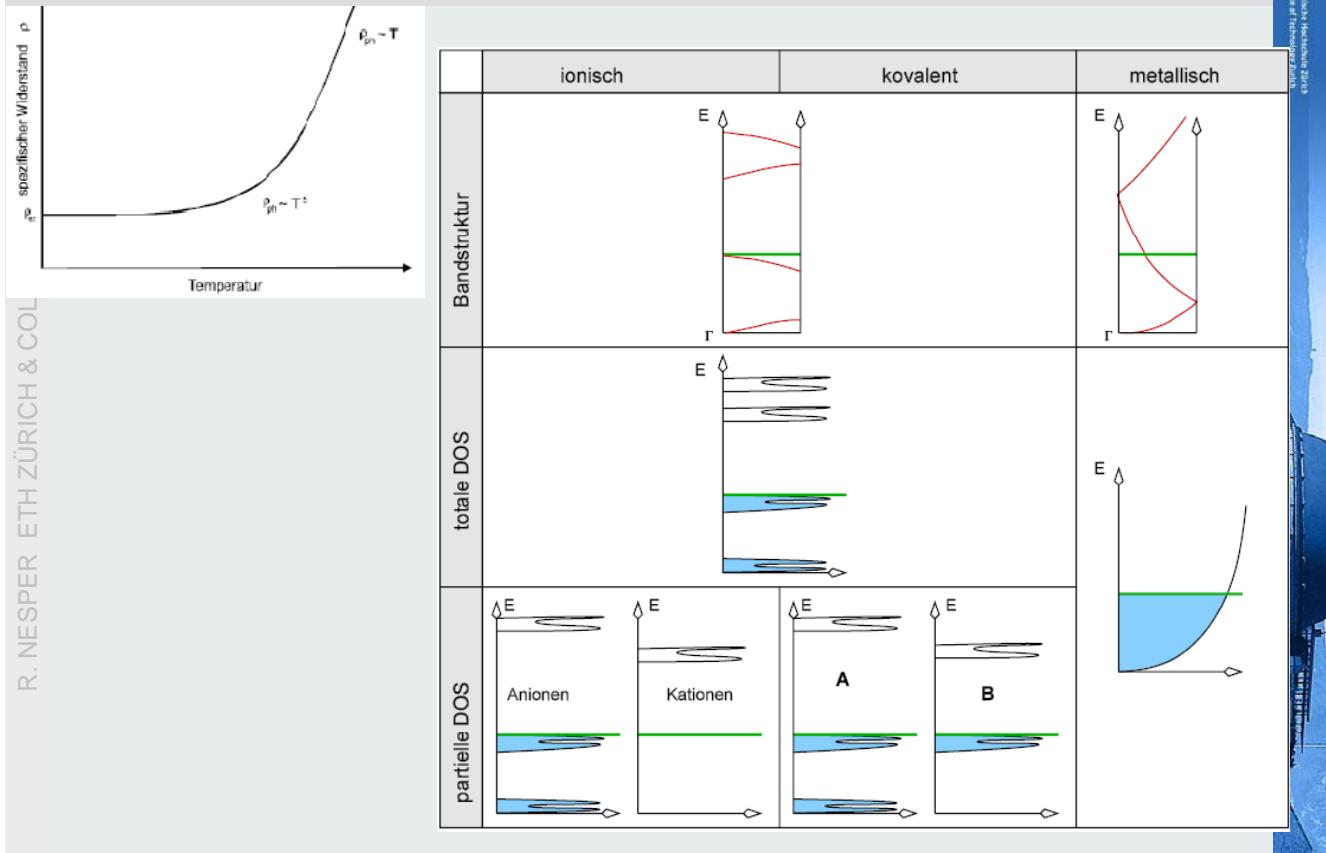


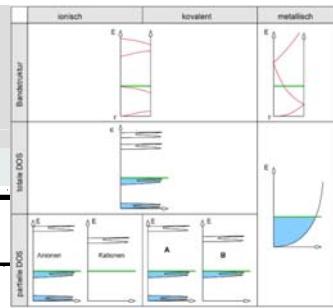
Metals, Semiconductors, Insulators



Metals, Semiconductors, Insulators

I-VII compounds	II-VI compounds	III-V compounds			
LiF	11	ZnO	3.4	AlP	3.0
LiCl	9.5	ZnS	3.8	AlAs	2.3
NaF	11.5	ZnSe	2.8	AlSb	1.5
NaCl	8.5	ZnTe	2.4	GaP	2.3
NaBr	7.5	CdO	2.3	GaAs	1.4
KF	11	CdS	2.45	GaSb	0.7
KCl	8.5	CdSe	1.8	InP	1.3
KBr	7.5	CdTe	1.45	InAs	0.3
KI	5.8	PbS	0.37	InSb	0.2
		PbSe	0.27	β -SiC	2.2
		PbTe	0.33	α -SiC	3.1

Some of these values, especially for the alkali halides, are only approximate.



Nitrides, Carbides ??

Nanochemistry UIO

Color in Semiconductors II Bandgap–Tuning by Quanten–Size–Effect

CdTe



Growing particle diametre from 2 auf 5 nm results in color change from green to red → bandgap becomes smaller

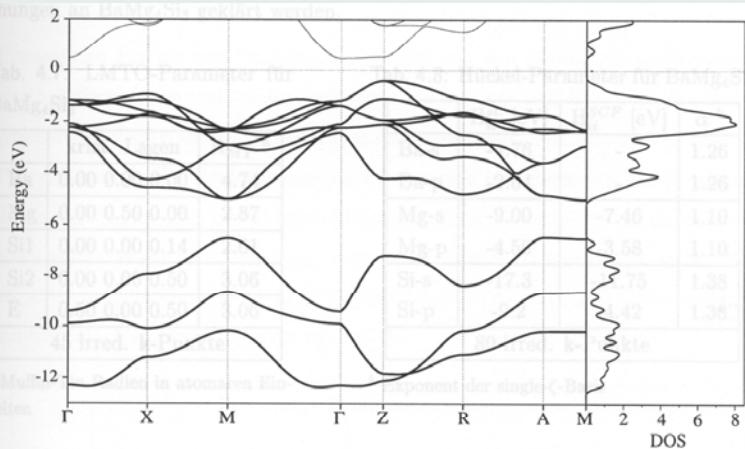
Nanochemistry UIO



Photoelectron spectra (ESCA)

UPS: $E_{\text{rad}} = \text{UV radiation} \Rightarrow$

valence states (bonds, etc)



binding
energy

Band structure

DOS

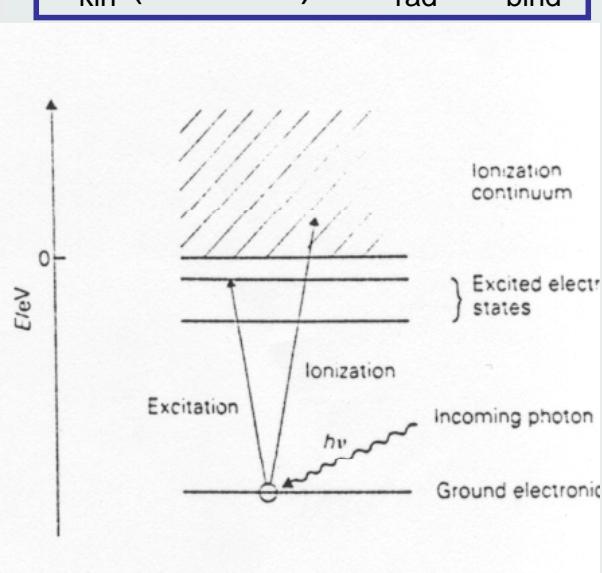
Nanochemistry UIO

Photoelectron spectra (ESCA)

$E_{\text{kin}} (\text{electron}) = E_{\text{rad}} - E_{\text{bind}}$

Surface method

2-3nm depth



Nanochemistry UIO

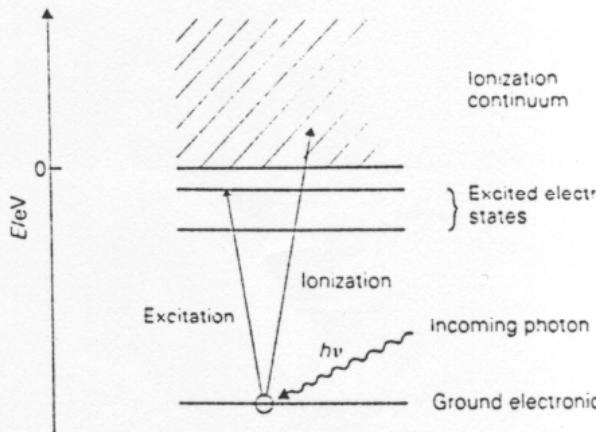
Photoelectron spectra (ESCA)

UPS: $E_{\text{rad}} = \text{UV radiation} \Rightarrow$

valence states (bonds, etc)

XPS: $E_{\text{rad}} = \text{X radiation} \Rightarrow$

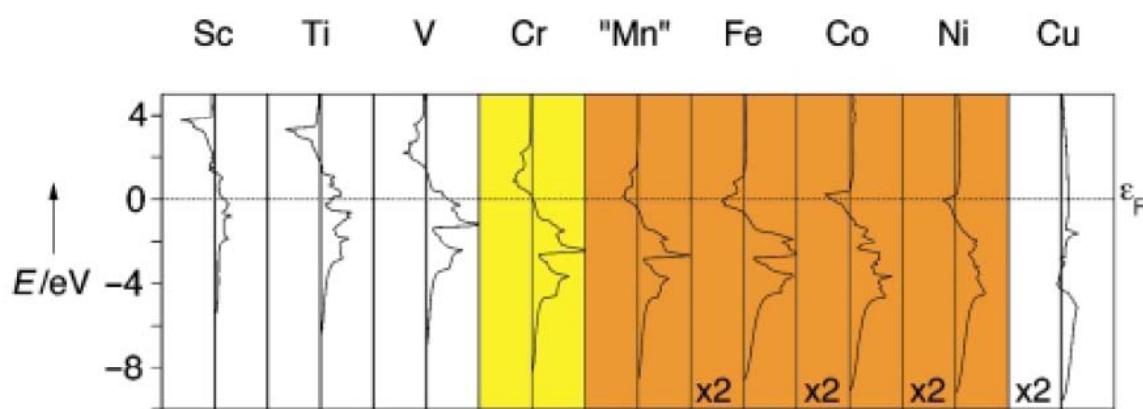
core states (oxidation states)



Nanochemistry UIO

Magnetism in Transition Metals

Angew. Chem. Int. Ed. **2000**, *39*, 1560–1585



Nanochemistry UIO

Magnetism - Spin Polarized Band Structures

Angew. Chem. Int. Ed. **2000**, *39*, 1560–1585

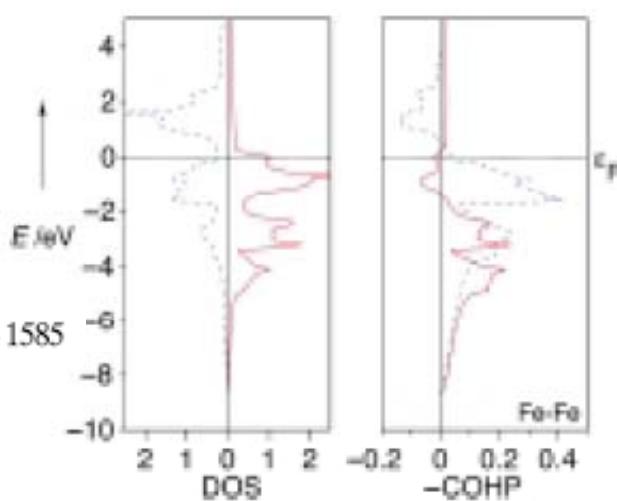


Figure 17. DOS and Fe-Fe COHP curves for ferromagnetic α -Fe. In each plot, the solid red/dashed blue line corresponds to the α/β spins. All curves have been shifted in energy so that ε_F , indicated with a horizontal dotted line, lies at 0.0 eV.

Nanochemistry UIO