Interaction theory –Electrons

Lesson FYSKJM4710

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Source: F. H. Attix: Introduction to radiological physics and radiation dosimetry (ISBN 0-471-01146-0)

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Excitation / ionization

• Incoming charged particle interacts with atom / molecule:



• An ion pair is created



Elastic collision 1

• Interaction between two particles where kinetic energy is preserved:



• Classical mechanics:

$$T_{0} = \frac{1}{2}m_{1}v^{2} = \frac{1}{2}m_{1}v_{1}^{2} + \frac{1}{2}m_{2}v_{2}^{2}$$
$$m_{1}v = m_{1}v_{1}\cos\theta + m_{2}v_{2}\cos\chi$$
$$0 = m_{1}v_{1}\sin\theta - m_{2}v_{2}\sin\chi$$

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Elastic collision 2

$$\Rightarrow \mathbf{v}_2 = \frac{2\mathbf{m}_1 \mathbf{v} \cos \chi}{\mathbf{m}_1 + \mathbf{m}_2} \quad , \quad \mathbf{v}_1 = \mathbf{v} \sqrt{1 - \frac{4\mathbf{m}_1 \mathbf{m}_2 \cos^2 \chi}{(\mathbf{m}_1 + \mathbf{m}_2)^2}}$$
$$\tan \theta = \frac{\sin 2\chi}{\frac{\mathbf{m}_1}{\mathbf{m}_2} - \cos 2\chi}$$

• Equations give, among others, maximum energy transferred:

$$E_{max} = \frac{1}{2}m_2v_{2,max}^2 = 4\frac{m_1m_2}{(m_1 + m_2)^2}T_0$$



Elastic collision 3

a) m ₁ >>m ₂	b) m ₁ =m ₂	c) m ₁ < <m<sub>2</m<sub>
$0 \le \chi \le \pi / 2$	$0 \le \chi \le \pi / 2$	$0 \le \chi \le \pi / 2$
$0 \le \theta \le \tan^{-1}(\frac{m_2}{m_1}\sin 2\chi)$	$0 \le \theta \le \pi / 2$	$0 \le \theta \le \pi$
$E_{max} = 4 \frac{m_2}{m_1} T_0$	$E_{max} = T_0$	$E_{max} = 4 \frac{m_1}{m_2} T_0$

• Proton-electron collision:

 $\theta_{max}=0.03^{o}$, $E_{max}=0.2$ %

• Electron-electron (or e.g. neutron-neutron) coll.:

 $\theta_{\text{max}} = 90^{\circ}$, $E_{\text{max}} = 100 \%$

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Elastic collision – cross section

• Rutherford showed that the cross section is:

$$\frac{d\sigma}{d\Omega} \propto \frac{1}{\sin^4(\theta/2)}$$

- \rightarrow small scattering angles are most probable
- With respect to energy:

$$\frac{d\sigma}{dE} \propto \frac{1}{E^2}$$

• \rightarrow small energy transfers most probable



Stopping power

• S=dT/dx; expected energy loss per unit lenght



$$dT = \left\langle En_{V}dx\sigma\right\rangle = n_{V}dx \int_{E_{min}}^{E_{max}} \frac{d\sigma}{dE} EdE = \rho\left(\frac{N_{A}Z}{A}\right) dx \int_{E_{min}}^{E_{max}} \frac{d\sigma}{dE} EdE$$
$$\left(\frac{dT}{\rho dx}\right) = \frac{S}{\rho} = \left(\frac{N_{A}Z}{A}\right) \int_{E_{min}}^{E_{max}} \frac{d\sigma}{dE} EdE$$

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Impact parameter

- Charged particles: Coulomb interactions
- Most important: interactions with electrons
- Impact parameter *b*:







Soft collisions 1

- b>> a : incoming particle passes atom at long distance
- Weak forces, small energy transfers to the atom
- Inelastic collisions: Predominantly excitations, some ionizations
- Energy transfer range from "*Emin*" to "*H*"
- Hans Bethe. Quantum mechanical considerations
- In the following, theory for heavy charged particles

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Soft collisions 2

$$\frac{S_{c,soft}}{\rho} = \left(\frac{dT_{soft}}{\rho dx}\right)_{c} = \frac{N_{A}Z}{A} \frac{2\pi r_{0}^{2}m_{e}c^{2}z^{2}}{\beta^{2}} \ln\left[\frac{2m_{e}c^{2}\beta^{2}H}{I^{2}(1-\beta^{2})} - \beta^{2}\right]$$

- r_0 : classical electron radius = $e^2/4\pi\epsilon_0 m_e c^2$
- I: mean excitation potential
- $\beta = v/c$
- z: charge of incoming particle
- ρ: density of medium
- N_AZ/A : numbers of electron per gram
- H: maksimum energy transferred by soft collisions



Soft collisions 2

• Quantum mechanics (atomic structure) is reflected in the mean excitation potential



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Hard collisions 1

- $b \ll a$: charged pass particle pass 'through' atom
- Large (but few) energy transfers
- Energy transfers from H to E_{max}
- May be considered as an elastic collision between free particles (binding energy is negligible)

$$\frac{S_{c,hard}}{\rho} = \left(\frac{dT_{hard}}{\rho dx}\right)_{c} = \frac{N_{A}Z}{A} \frac{2\pi r_{0}^{2}m_{e}c^{2}z^{2}}{\beta^{2}} \ln\left[\frac{E_{max}}{H} - \beta^{2}\right]$$



Collision stopping power

• For inelastic collisions, the total cross section is thus:

$$\frac{S_{c}}{\rho} = \frac{S_{c,soft}}{\rho} + \frac{S_{c,hard}}{\rho}$$
$$= 4\pi r_{0}^{2} m_{e} c^{2} \left(\frac{N_{A}Z}{A}\right) \left(\frac{z}{\beta}\right)^{2} \left[\ln\left(\frac{2m_{e}c^{2}\beta^{2}}{(1-\beta^{2})I}\right) - \beta^{2}\right]$$

• Important: Increases with z², decreases with v², not dependent on particle mass

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S_c/ρ , different substances



• I and electron density (ZN_A/A) give differences





S_c/ρ , electrons and positrons

- Electron-electron scattering is more complicated; scattering between two identical particles
- $S_{c, hard}/\rho$ (el-el) is described by the Möller cross section
- $S_{c, hard}/\rho$ (pos-el) is described by the Bhabha c.s.
- $S_{c, soft} / \rho$ was given by Bethe, as for heavy particles
- Characteristics similar to that for heavy charged particles

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Shell correction

- Derivation of S_c assumes v >> v_{atomic electrons}
- When $v \sim v_{atomic \ electrons}$, no ionizations
- Most important for K-shell electrons
- Shell correction C/Z takes this into accout, and thus reduces S_c/ρ
- C/Z depends on particle energy and medium





Density correction

• Charged particles polarizes medium which is being traversed



- Weaker interactions with remote atoms due to reduction in electromagnetic field strenght
- Polarization increases with energy and density
- Most important for electrons and positrons

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Density correction

- Density correction δ reduces S_c/ρ for liquids and solids
- S_c/ρ (water vapor) > S_c/ρ (water)



Dashed line: S_c/ρ without δ



Linear Energy Transfer 1

- LET_{Δ} is denoted the *restricted stopping power*
- dT/dx: mean energy loss per unit lenght but how much is deposited 'locally'?



Electrons leaving 'local volume' \rightarrow energy transfer $> \Delta$

- S_c : energy transfers from E_{min} to E_{max}
- How much energy is deposited within the range of an electron given energy Δ ?

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Linear Energy Transfer 2

• Energy loss (soft + hard) per unit lenght for $E_{min} < E < \Delta$:

$$L_{\Delta} = \left(\frac{dT}{dx}\right)_{\Delta} = \rho \left(\frac{N_{A}Z}{A}\right)_{E_{min}}^{\Delta} \frac{d\sigma}{dE} E dE$$
$$= \rho 2\pi r_{0}^{2} m_{e} c^{2} \left(\frac{N_{A}Z}{A}\right) \left(\frac{z}{\beta}\right)^{2} \left[\ln\left(\frac{2m_{e}c^{2}\beta^{2}\Delta}{(1-\beta^{2})I}\right) - 2\beta^{2}\right]$$

- For $\Delta = E_{max}$, we have $L_{\infty} = S_c$; unrestricted LET
- LET_{Δ} is often given in [keV/ μ m]
- 30 MeV protons in water: LET_{100 eV} / $L_{\infty} = 0.53$



Brehmsstrahlung 1

• Photon may be emitted from charged particle accelerated in the field from an electron or nucleus



• Larmor's formula (classical electromagnetism) for radiated effect from accelerated charged particle:

$$P = \frac{(ze)^2 a^2}{6\pi\epsilon_0 c^3}$$

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Brehmsstrahlung 2

• For particle accelerated in nuclear field:

$$F = ma = \frac{zZe^2}{4\pi\epsilon_0 r^2} \implies a = \frac{zZe^2}{4\pi\epsilon_0 mr^2}$$
$$\implies P \propto \left(\frac{Z}{m}\right)^2$$

• Comparison of protons and electrons:

$$\frac{P_{p}}{P_{e}} = \left(\frac{m_{e}}{m_{p}}\right)^{2} \approx \frac{1}{1836^{2}}$$

• Brehmsstrahlung not important for heavy charged particles







Brehmsstrahlung 3

- Energy loss by brehmsstrahlung is called *radiative loss*
- Maksimum energy loss is the totale kinetic energy T
- Radiative loss per unit lenght: *radiative stopping power*:

$$\left(\frac{S}{\rho}\right)_{r} = \left(\frac{dT}{\rho dx}\right)_{r} \approx \alpha r_{0}^{2} \frac{N_{A}Z^{2}}{A} (T + m_{e}c^{2})\overline{B_{r}}(T, Z)$$

- $\overline{B_r}(T,Z)$ weakly dependent on T and Z
- Brehmsstrahlung increases with energy and atomic number

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Total stopping power, electrons

• Total mass stopping power:







Radiation yield



 S_c/ρ , protons and electrons







Cerenkov effect

- High energy electrons (v > c/n) polarizes medium (e.g. water) and blueish light (+ UV) is emitted
- Low energy loss



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Other interactions

- *Nuclear interactions*: Inelastic process where charged particle (e.g. proton) excites nucleus →
 - Scattering of charged particle
 - Emission of neutron, photon, or α -particle (⁴₂He)
- Not important below ~10 MeV (protons)
- Positron annihilation: Positron interacts with electron → a pair of photons with energy ≥ 2 x 0.511 MeV is created. Photons are emitted in opposite directions.
- Probability decreases as $\sim 1/v$



Range 1

- The range \mathcal{R} of a charged particle in matter is the (expectation value) of it's total pathlenght p
- The projected range $\langle t \rangle$ er is the (expectation value) of the largest depth t_f a charged particle can reach along it's incident direction
- Electrons: $< t > < \Re$
- Heavy charged particles: $\langle t \rangle \approx \Re$



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CSDA-range

- The range may be approximated by \mathcal{R}_{CSDA} (continuous slowing down approximation)
- Energy loss per unit lenght dT/dx gives implicitly a measure of the range:

$$\overrightarrow{T_{0}} \xrightarrow{T_{0} - \Delta T} = T_{0} - \frac{dT}{dx} \Delta x$$
$$\Delta x = \frac{dx}{dT} \Delta T \quad , \quad \Rightarrow \Re = \sum_{i=1}^{n} \Delta x_{i} = \sum_{i=1}^{n} \left(\frac{dx}{dT}\right)_{i} \Delta T$$
$$\Rightarrow \Re_{CSDA} = \int_{0}^{T_{0}} \left(\frac{dT}{dx}\right)^{-1} dT$$





Range 3

• The range is often given multiplied by the density:

$$\Re_{\rm CSDA} = \int_0^{T_0} \left(\frac{dT}{\rho dx}\right)^{-1} dT$$

- Unit thus becomes $[cm] [g/cm^3] = [g/cm^2]$
- Range of charged particle depends on:
 - Charge and kinetic energy
 - Density, electron density and mean excitation potential of absorber





Multiple scattering and straggling

- In a beam of charged particles, one has:
 - Variations in energy deposition (straggling)
 - Variations in angular scattering
- → The beam, where all particles originally had the same velocity, will be smeared out as the particles traverses matter



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Straggling









Energy deposition, electrons



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Monte Carlo simulations

- Monte Carlo simulations of the track of an electron (0.5 keV) and an α-particle (4 MeV) in water
- Note:

 e^{-} is most scattered α has the highest dT/dx



Hadron therapy

• Heavy charged particles may be used for radiation therapy – conforms better to the target than photons or electrons



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Web pages

• For stopping powers:

http://physics.nist.gov/PhysRefData/Star/Text/contents.html

• For attenuation coefficients:

http://physics.nist.gov/PhysRefData/XrayMassCoef/cover.html

