

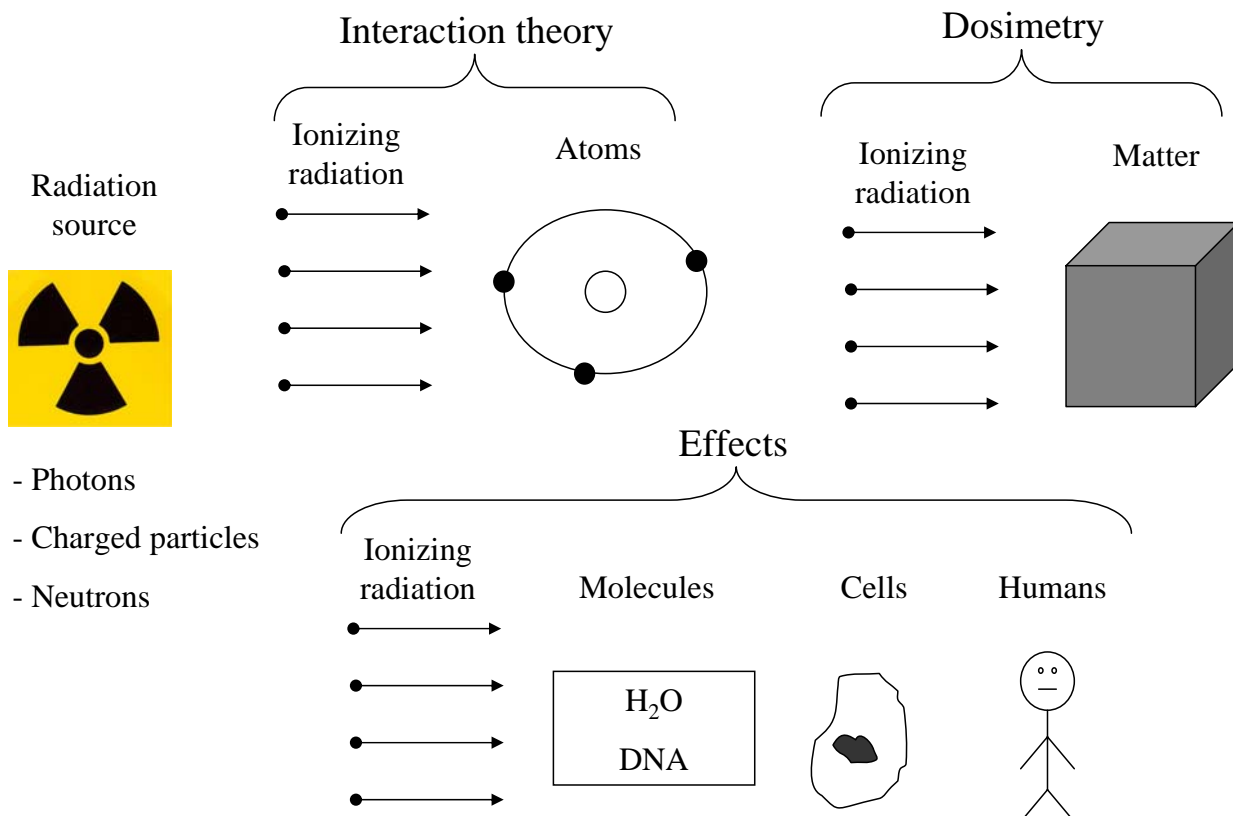
# Interaction theory – Photons

## Lesson FYSKJM4710

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Source: F. H. Attix: Introduction to radiological physics and radiation dosimetry (ISBN 0-471-01146-0)

## Introduction



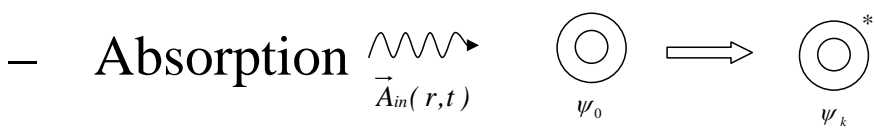
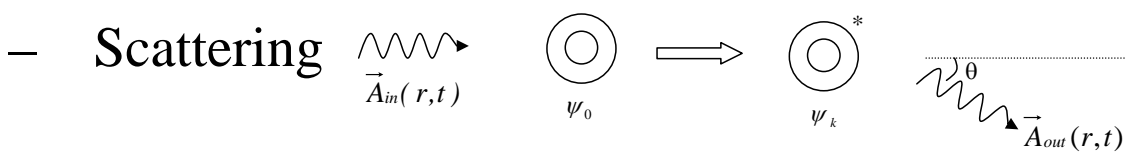
# Contents FYSKJM4710

- Interactions between ionizing radiation and matter
- Radioactive and non-radioactive sources
- Calculations and measurement of absorbed doses (dosimetry)
- Radiation chemistry
- Biological effects of ionizing radiation
- Principles of radiation protection

## Objectives

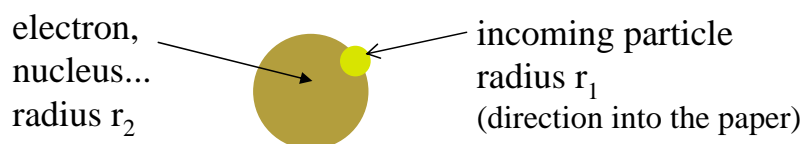
- To understand primary and secondary effects of ionizing radiation
- How radiation doses are calculated and measured
- To understand the principles of radiation protection, their origin and applications

# Photon interactions

- Photon represented by a plane wave  $\vec{A}_{in}(r,t) \sim e^{i(\vec{p}_{in} \cdot \vec{r} - \omega_{in} t)}$  in quantum mechanical calculations
- In principle, two different processes:
  - Absorption 
  - Scattering 
- Scattering: coherent (elastic) og incoherent (inelastic)

## Cross section 1

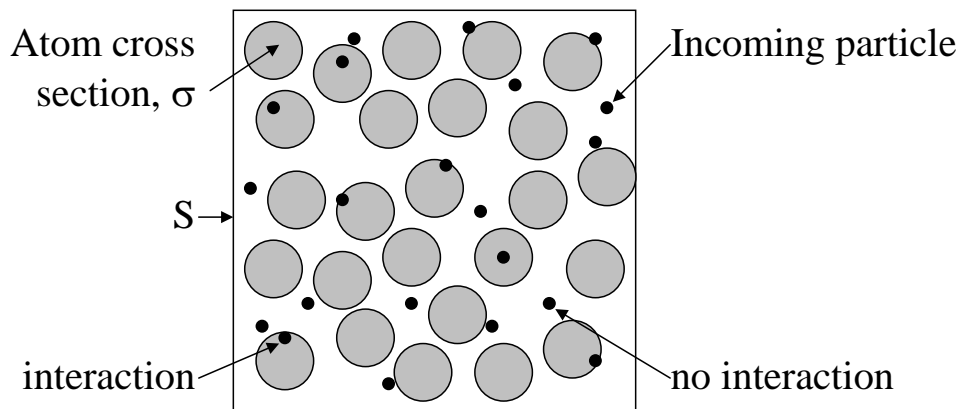
- Cross section  $\sigma$ : “target area”, effective target covering a certain area
- Proportional to the interaction strength between an incoming particle and the target particle
- Consider two discs, one target and one incoming:



- $\sigma$  is the total area:  $\pi(r_1^2 + r_2^2)$

# Cross section 2

- $N$  particles move towards an area  $S$  with  $n$  atoms



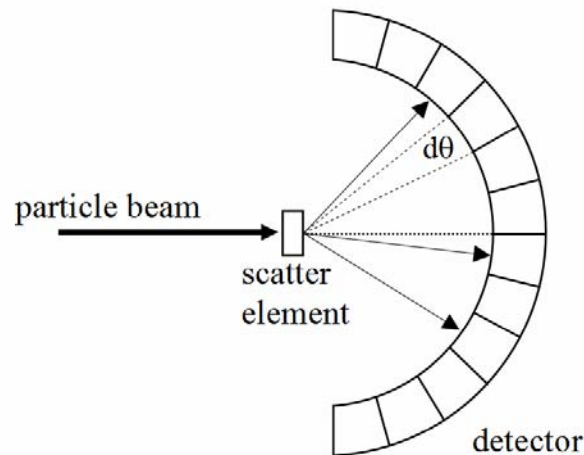
- Probability of interaction:  $p = n\sigma/S$
- Number of interacting particles:  $Np = Nn\sigma/S$

# Cross section 3

- Separate between *electronic* and *atomic* cross section
- The cross section depends on:
  - Type of target (nucleus, electron, ..)
  - Type of and energy of incoming particle (photon, electron...)
- Cross section calculated with quantum mechanics
  - here visualized in a classical window

# Cross section 4

- *Differential cross section* with respect to scattering angle



$$\frac{d\sigma}{d\Omega} = \frac{\text{number of particles scattered into } d\Omega}{\text{number of particles per unit area}} \frac{1}{d\Omega}$$

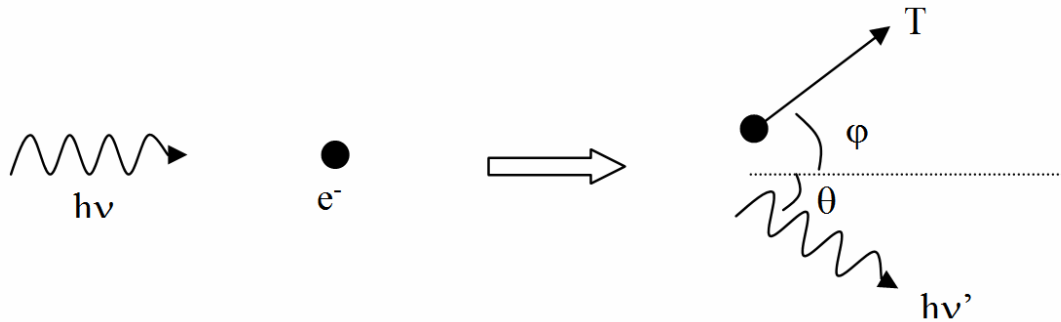
# Coherent (Rayleigh) scattering

- Scattering without loss of energy:  $h\nu = h\nu'$
- Photon is absorbed by atom, thereby emitted at a small deflection angle
- Depends on atomic structure and photon energy
- Atomic cross section:

$$\sigma_R \propto \left( \frac{Z}{h\nu} \right)^2$$

# Incoherent (Compton) scattering

- Scattering with loss of energy:  $h\nu' < h\nu$
- Photon-electron scattering; electron may be assumed free (i.e. unbound)



## Compton scattering – kinematics

- Conservation of energy and momentum:

$$h\nu = h\nu' + T$$

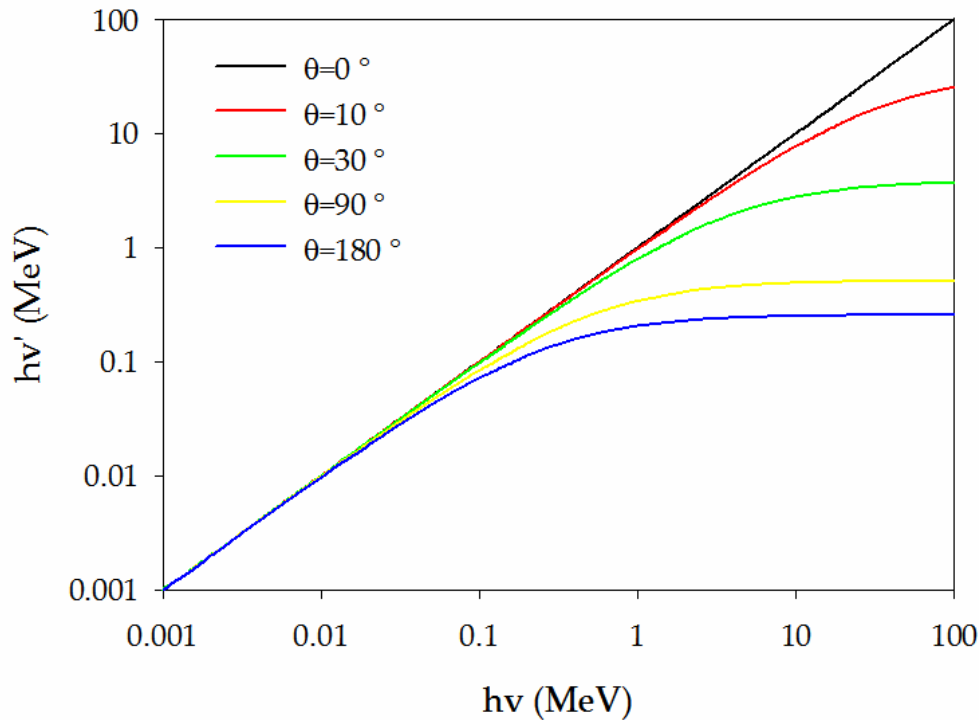
$$\frac{h\nu}{c} = \frac{h\nu'}{c} \cos \theta + p \cos \varphi \quad , \quad \frac{h\nu'}{c} \sin \theta = p \sin \varphi$$

$$(pc)^2 = T^2 + 2Tm_e c^2$$

→

$$h\nu' = \frac{h\nu}{1 + \frac{h\nu}{m_e c^2} (1 - \cos \theta)} \quad , \quad \cot \varphi = \left( 1 + \frac{h\nu}{m_e c^2} \right) \tan \left( \frac{\theta}{2} \right)$$

# Compton scattering – kinematics



## Compton scattering – example

- An X-ray unit is to be installed, with the beam direction towards the ground. Employees in the floor above the unit are worried. Maximum X-ray energy is 250 keV. What is the maximum energy of the backscattered photons?

$$\theta = 180^\circ \Rightarrow hv' = \frac{hv}{1 + \frac{hv}{m_e c^2} (1 - \cos \theta)} = \frac{hv}{1 + \frac{2hv}{m_e c^2}}$$

$$hv = 250 \text{ keV} \Rightarrow hv' = \frac{250}{1 + \frac{2 \times 250}{511}} = \underline{\underline{126 \text{ keV}}}$$

# Compton scattering – cross section 1

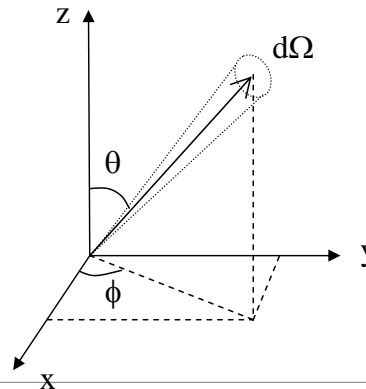
- Klein and Nishina derived the cross section for Compton scattering, assuming free electron
- Differential cross section:

$$\left( \frac{d\sigma}{d\Omega} \right) = \frac{r_0^2}{2} \left( \frac{\nu'}{\nu} \right)^2 \left( \frac{\nu'}{\nu} + \frac{\nu}{\nu'} - \sin^2 \theta \right)$$

$$d\Omega = \sin \theta d\theta d\phi$$

$r_0$ : classical electron radius

incoming photon along z-axis



# Compton scattering – cross section 2

- Cylinder symmetry results in:

$$\left( \frac{d\sigma}{d\theta} \right) = \pi r_0^2 \left( \frac{\nu'}{\nu} \right)^2 \left( \frac{\nu'}{\nu} + \frac{\nu}{\nu'} - \sin^2 \theta \right) \sin \theta$$

- ~ probability of finding a scattered photon in the interval  $[\theta, \theta+d\theta]$
- Total electronic cross section:

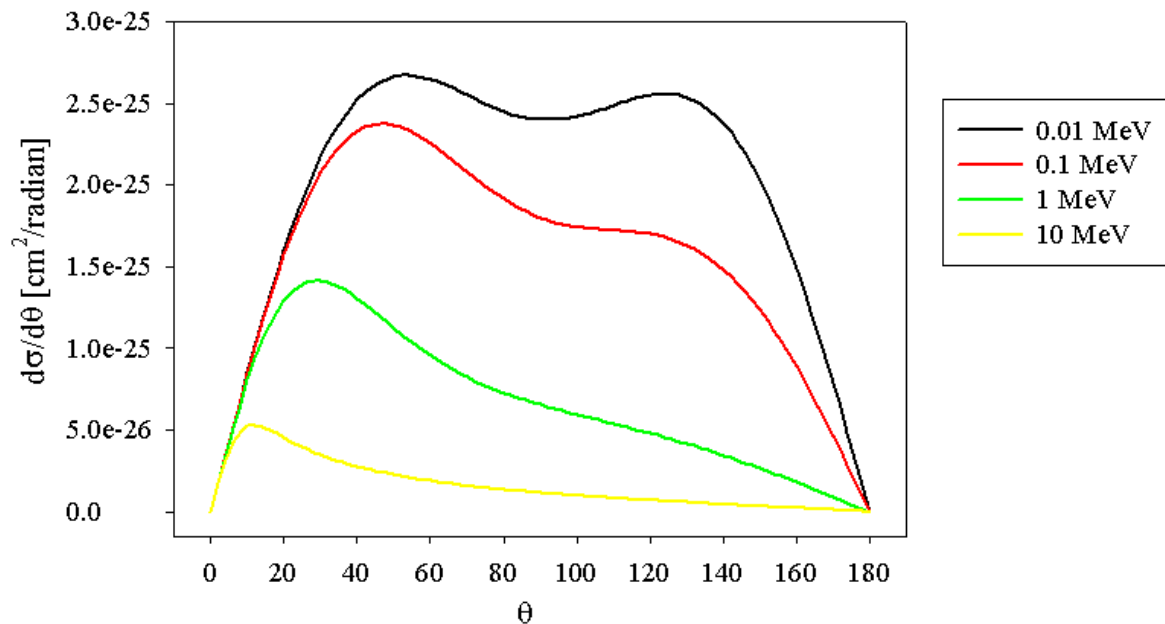
$${}_e\sigma = \int_0^\pi \pi r_0^2 \left( \frac{\nu'}{\nu} \right)^2 \left( \frac{\nu'}{\nu} + \frac{\nu}{\nu'} - \sin^2 \theta \right) \sin \theta d\theta$$

- Atomic cross section:  ${}_a\sigma = Z {}_e\sigma$



# Compton scattering – cross section 3

- Scattered photons are more forwardly directed with increasing photon energy:



# Compton scattering – cross section 3

- Cross section may be modified with respect to energy:

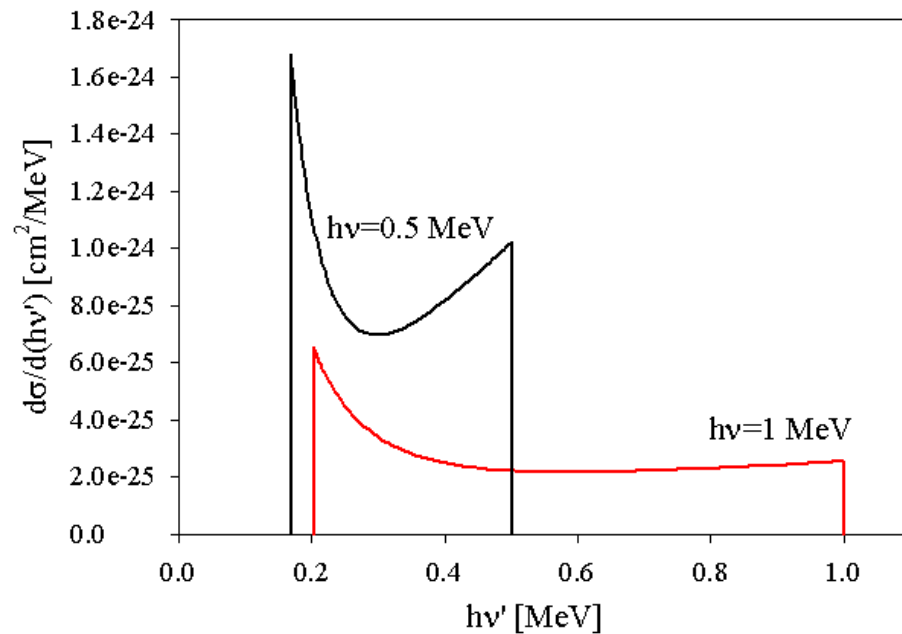
$$\frac{d\sigma}{d(h\nu')} = \frac{d\sigma}{d\Omega} \frac{d\Omega}{d(h\nu')} = \frac{d\sigma}{d\Omega} 2\pi \sin\theta \frac{d\theta}{d(h\nu')}$$

$$h\nu' = \frac{h\nu}{1 + \frac{h\nu}{m_e c^2} (1 - \cos\theta)}$$

$$\Rightarrow \frac{d\sigma}{d(h\nu')} = \frac{\pi r_0^2 m_e c^2}{(h\nu)^2} \left[ \frac{h\nu'}{h\nu} + \frac{h\nu}{h\nu'} - 1 + \left( 1 - \left( \frac{h\nu}{h\nu'} - 1 \right) \frac{m_e c^2}{h\nu} \right)^2 \right]$$

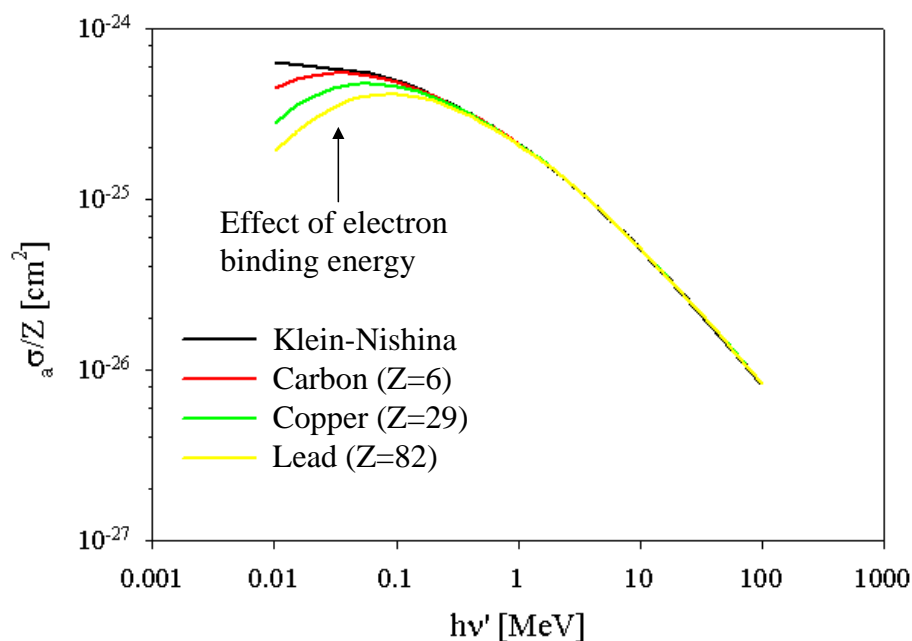
# Compton scattering – cross section 4

- Cross section may be modified with respect to energy:



# Compton scattering – cross section 5

- Correct atomic cross section:



# Compton scattering – transferred energy 1

- The energy transferred to an electron in a Compton process:

$$T = h\nu - h\nu'$$

- The cross section for energy transfer:

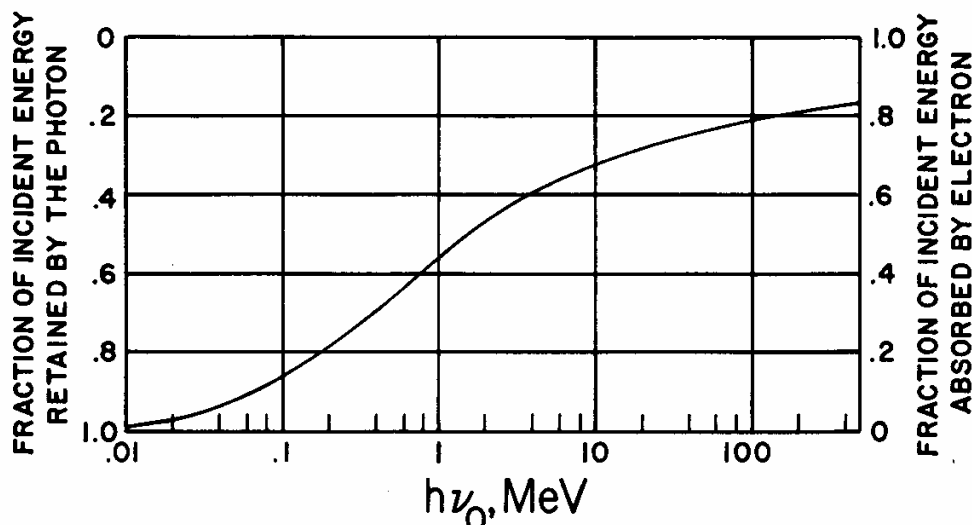
$$\frac{d\sigma_{tr}}{d\Omega} = \frac{d\sigma}{d\Omega} \frac{T}{h\nu} = \frac{d\sigma}{d\Omega} \frac{h\nu - h\nu'}{h\nu}$$

- Mean energy transferred:

$$\bar{T} = \frac{\int T \frac{d\sigma}{d\Omega} d\Omega}{\int \frac{d\sigma}{d\Omega} d\Omega} = \frac{\int \frac{h\nu - h\nu'}{h\nu} \frac{d\sigma}{d\Omega} d\Omega}{\sigma} = \frac{\sigma_{tr}}{\sigma}$$

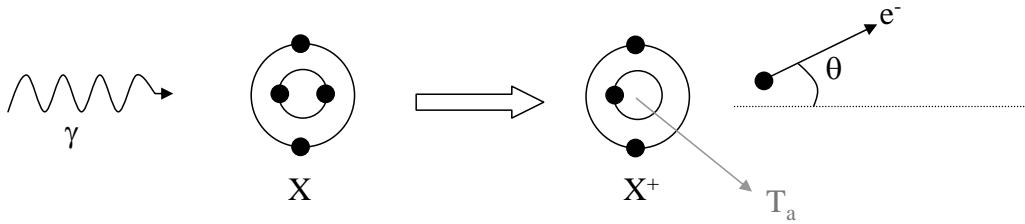
# Compton scattering – transferred energy 2

- The fraction of incident energy transferred:

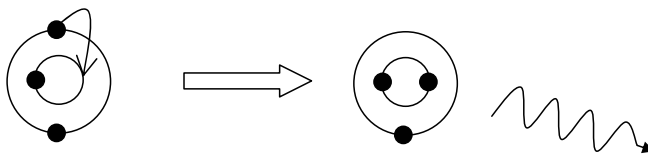


# Photoelectric effect 1

- Photon is absorbed by atom/molecule; the result is an excitation or ionization



- Atom may deexcite and emit characteristic radiation:



# Photoelectric effect 2

- In the kinematics, the binding energy of the ejected electron should be taken into account:

$$T = h\nu - E_b - T_a \approx h\nu - E_b$$

- Assuming  $E_b=0$ , the atomic cross section is:

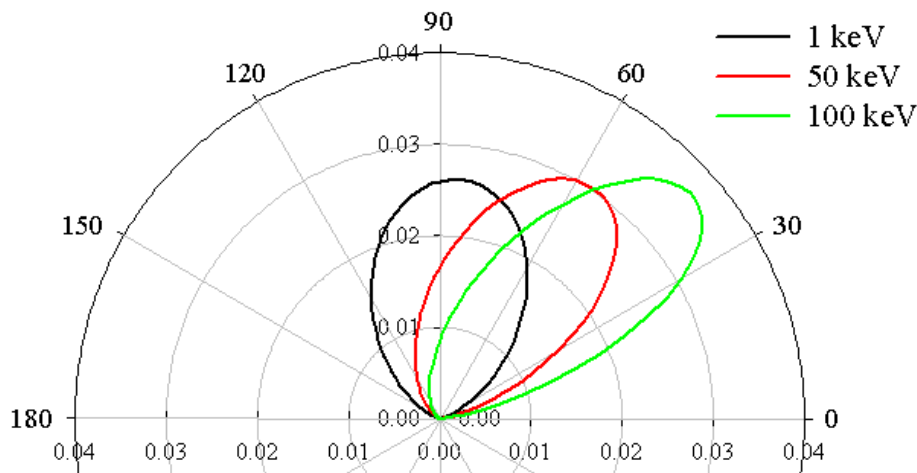
$$\frac{d\tau}{d_e\Omega} = 2\sqrt{2}r_0^2\alpha^4Z^5\left(\frac{m_e c^2}{h\nu}\right)^{7/2}\sin^2\theta\left(1+4\sqrt{\frac{2h\nu}{m_e c^2}}\cos\theta\right)$$

$\alpha$ : The fine-structure constant

Solid angle  $d_e\Omega$  gives the direction of the ejected electron

# Photoelectric effect 2

Photoelectric cross section  $(d\sigma/d\theta)/\sigma$



## Characteristic radiation

- Energy of characteristic radiation depends on electronic structure and transition probabilities
- "K- and L-shell" vacancies  $\leftrightarrow h\nu_K$  and  $h\nu_L$
- Isotropic emission
- Fraction of photoelectric interactions:  
 $P_K [h\nu > (E_b)_K]$  and  $P_L [(E_b)_K < h\nu < (E_b)_K]$
- Probability for emission:  $Y_K$  og  $Y_L$  (fluorescence yield)

- Energy emitted from the atom:

$$P_K Y_K h\nu_K + (1 - P_K) P_L Y_L h\nu_L$$

# Auger effect

- Energy release by ejection of loosely bound electron
- Energy of emitted electron equal to deexcitation energy
- Low Z: Auger dominates
- High Z: characteristic radiation dominates

# Photoelectric cross section

- General formula:

$$\tau \propto \frac{Z^n}{(h\nu)^m}, \quad 4 < n < 5, \quad 1 < m < 3$$

- Fraction of energy transferred to photoelectron:

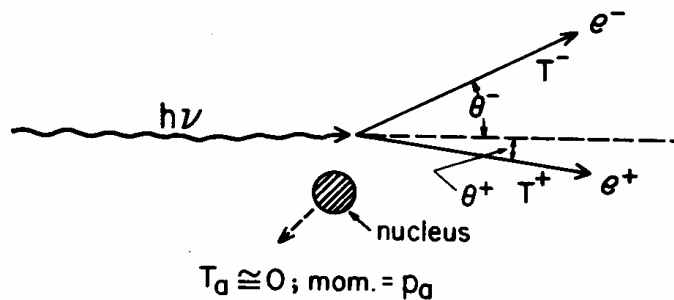
$$\frac{T}{h\nu} = \frac{h\nu - E_b}{h\nu}$$

- However: don't forget Auger electron(s)
- Cross section for energy transfer to photoelectron:

$$\tau_{tr} = \tau \frac{(h\nu - P_K Y_K h\nu_K - (1 - P_K) P_L Y_L h\nu_L)}{h\nu}$$

# Pair production 1

- Photon absorption in the nuclear electromagnetic field where an electron-positron pair is created



- Triplet production: in the electromagnetic field of an electron

# Pair production 2

- Conservation of energy:

$$h\nu = 2m_e c^2 + T^+ + T^-$$

- Average kinetic energy after absorption:

$$\bar{T} = \frac{h\nu - 2m_e c^2}{2}$$

- Estimated electron/positron scattering angle:

$$\bar{\theta} \approx \frac{m_e c^2}{\bar{T}}$$

- Total cross section:

$$\kappa \approx \alpha r_0^2 Z^2 \bar{P}$$

# Triplet production

- In the electromagnetic field from an electron, an electron-positron pair is created

- Energy conservation:

$$h\nu = 2m_e c^2 + T^+ + T_1^- + T_2^-$$

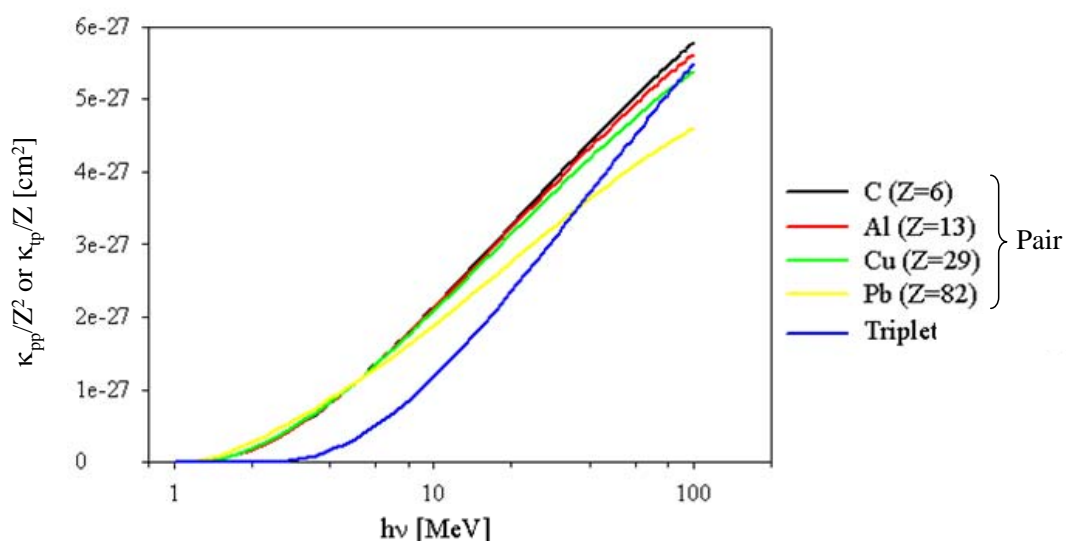
- Average kinetic energy:

$$\bar{T} = \frac{h\nu - 2m_e c^2}{3}$$

- Primary electron is also given energy
- Threshold:  $4m_0c^2$

# Pair- and triplet production

- Pair production dominates:



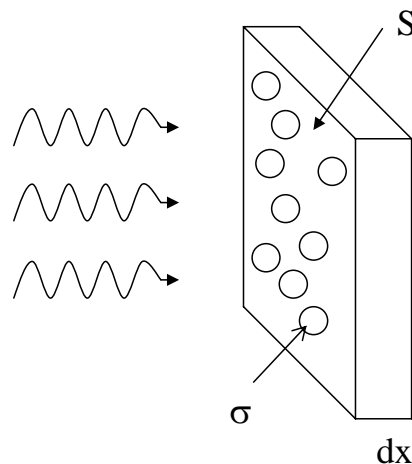


# Photonuclear reactions

- Photon (energy above a few MeV) excites a nucleus
- Proton or neutron is emitted
- $(\gamma, n)$  interactions may have consequences for radiation protection
- Example: Tungsten W  $(\gamma, n)$

## Attenuation coefficients 1

- $n_V$  atoms per volume =  $\rho(N_A/A)$
- Number of atoms:  
 $n = n_V V = n_V S dx$
- Interaction probability  
 $p = n\sigma/S = n_V \sigma dx$
- Probability per unit length:  
 $\mu = p/dx = n_V \sigma = \rho(N_A/A)\sigma$   
 $\mu$ : linear attenuation coefficient



## Attenuation coefficients 2

- $N_A$  : Avogadro's constant;  $6.022 \times 10^{23} \text{ mole}^{-1}$
- $A$ : number of grams per mole
- $N_A/A$ : number of atoms per gram
- $N_A Z/A$ : number of electrons per gram
- Number of atoms per volume:  $\rho(N_A/A)$
- Etc.

## Attenuation coefficients 3

- Total *mass* attenuation coefficient:

$$\frac{\mu}{\rho} = \frac{\tau}{\rho} + \frac{\sigma}{\rho} + \frac{\kappa}{\rho} + \frac{\sigma_R}{\rho}$$

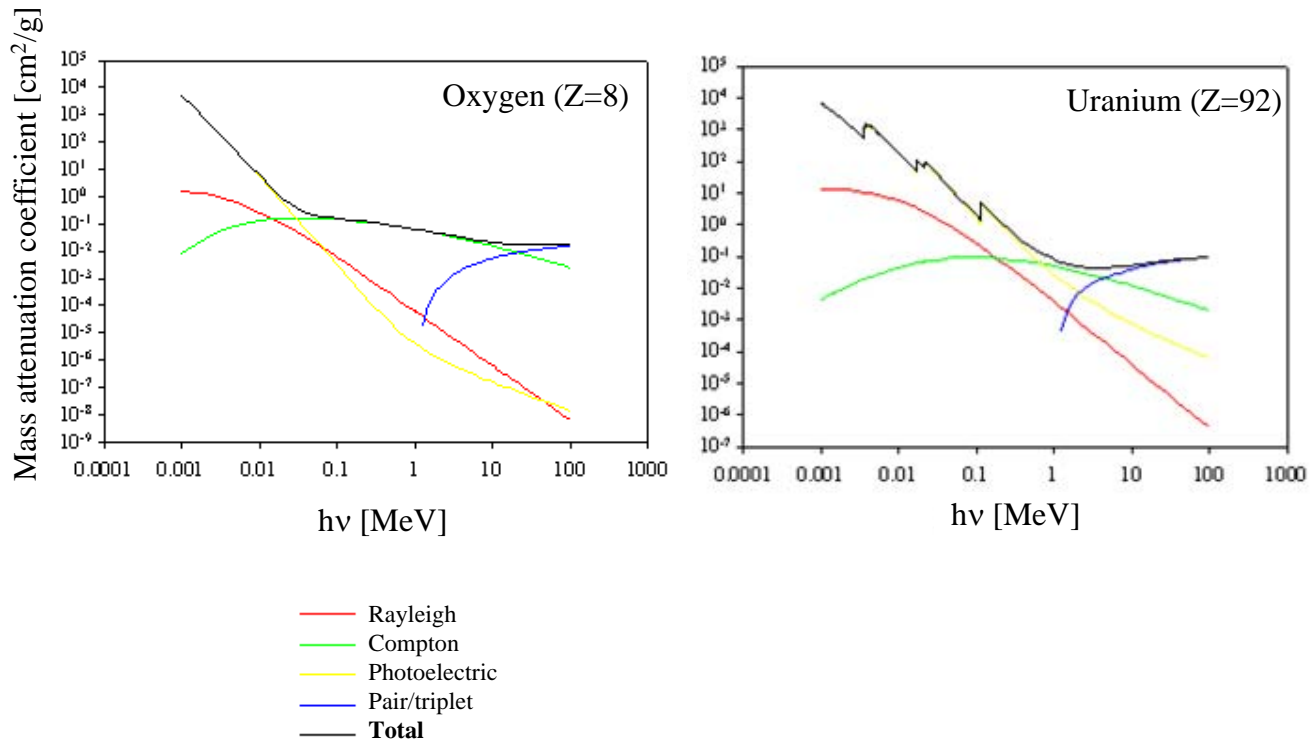
- Coefficient for energy transfer:

$$\frac{\mu_{tr}}{\rho} = \frac{\mu}{\rho} \frac{\bar{T}}{h\nu}$$

- Braggs rule for mixture of atoms:

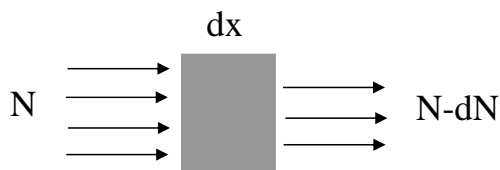
$$\left(\frac{\mu}{\rho}\right)_{mix} = \sum_{i=1}^n f_i \left(\frac{\mu}{\rho}\right)_i, \quad f_i = \frac{m_i}{\sum_{i=1}^n m_i}$$

# Attenuation coefficients 4



## Attenuation

- Beam with  $N$  photons impinge absorber with thickness  $dx$ :



- Sannsynlighet for at ett foton skal vekselvirke:  $\mu dx$
- Antall fotoner som vekselvirker:  $N\mu dx$

$$dN = N\mu dx \quad \Rightarrow \quad \int \frac{dN}{N} = \int \mu dx$$

$$\Rightarrow \underline{\underline{N = N_0 e^{-\mu x}}}$$

# Mean free path

- 'Probability' for photon not interacting:  $e^{-\mu x}$
- Normalized probability

$$p_{ni} = Ce^{-\mu x}, \int_0^{\infty} p_{ni} dx = 1, \Rightarrow p_{ni} = \mu e^{-\mu x}$$

- Mean free path:

$$\langle x \rangle = \int_0^{\infty} x p_{ni} dx = \int_0^{\infty} x \mu e^{-\mu x} dx = \frac{1}{\mu}$$

## Summary

